Chem 130: Chemistry for Funeral Services Problem Set 10

Name:	KEY	Date:
Each q	question is worth one point. Show your work whereve	er calculations are required.
	Describe in some detail the characteristics of one major on oose either proteins, carbohydrates, fats or nucleic acherwise) and use in biological systems.	
2.	Why is peptide bond formation considered to be a neutra	lization reaction?
	A peptide bond is simply an amide bond. Amide bon amines. Carboxylic acids are organic acids. Amines in biological systems—form in acid base reactions.	are organic bases. So amides—peptide bonds
3.	Show how formaldehyde cross-links protein molecules.	How does cross-linking by gluteraldehyde differ?
	Formaldehyde forms cross-links wherever the —NH—the imide group in some amino acid side chains, the NH— found in peptide bonds. Formaldehyde reacts NH from another of these sources to insert a methyle This can happen between each peptide bond in a progroups between its two aldehyde groups so it cannot cross-link every other amide nitrogen. See diagrams	terminal amine group in proteins and the —with the NH from one of these sources and the ene group —CH ₂ — between the two nitrogens. Instead, it can be to crosslink every amide nitrogen. Instead, it can
4.	What is an amphoteric substance? Give an important ex	cample related to proteins.
	An amphoteric substance can act either as an acid o	·
E	Describe what happens when the hamaglahin male wile h	stacks down. How does this affect amb almin of
5.	Describe what happens when the hemoglobin molecule b	•
	The heme group is released from the protein structur opens up creating a long chain with carbons, nitroge (green color) is the first important product. It can be Hemoglobin release to tissues causes postmortem s in jaundice. Both are cosmetic issues faced in emba	ns and alternating double bonds. Biliverdin further broken down to bilirubin (yellow color). tain. Bilirubin causes the yellow coloring seen
6.	How does the secondary structure of proteins relate to his discussed in class and show how hydrogen bonding is in	
	Hydrogen bonding is one of the primary binding force Depending on which atoms are hydrogen bonded three get helixes or pleats. Further description of alpha he class notes and the text.	oughout parts of the protein structure, you can
7.	How is denaturation of a protein different from protein brownen a protein denatures it loses much or all of its sits function (depending on the extent of the disruption results in the formation of amino acids by hydrolysis peptide chain. Disinfecting agents are used in embasis.	econdary and tertiary structure. It usually loses n to the protein structure.) Protein breakdown of the primary structure of the protein – its

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8. Describe key characteristics of enzymes and their role in biological processes. What is the lock and key model?

Enzymes are biological catalysts. The lock and key model is one way of describing how the substrate fits into an active site. By this model, if the substrate doesn't fit exactly, much like a key, the enzyme won't work on it. The enzyme remains "locked". A more accurate model is to describe the process as "induced fit". The substrate fits into the active site and the enzyme shifts slightly to mold itself to the substrate. This model is more like a hand (substrate) fitting into a glove (enzyme).

9. Describe how enzymes affect the decomposition process.

The decomposition process is governed by the action of a variety of enzymes. Different enzymes are responsible for parts of the putrefaction process—proteolytic enzymes that breakdown proteins and hydrolytic enzymes that govern a wide variety of reactions involving the breakdown of substances through the addition of water to bonds.

10. The chart below includes four of the processes involved in protein breakdown. Complete the chart.

Type of process	What happens?	How it affects embalming:
Digestion	One way of beginning of protein breakdown. Peptide chains are broken into a series of large segments.	More "N terminus" nitrogens are available because of the shorter chains. This causes more amine groups to be available to use formaldehyde.
Hydrolysis	Proteins are broken down into amino acids by the addition of water to the peptide (amide) bonds	There are free amine groups on every amino acid. Formaldehyde demand increases.
Deamination	Amino acids lose their amine group. Ammonia and carboxylic acids are produced.	Ammonia uses up formaldehyde. Formaldehyde demand increases.
Decarboxylation	Amino acids lose their carboxylic acid groups. Carbon dioxide, water and an amine result.	Amine groups can react with formaldehyde. They can also be further broken down to release ammonia. Either way, formaldehyde demand increases.